

```
=> fil caplus
COST IN U.S. DOLLARS          SINCE FILE      TOTAL
                               ENTRY      SESSION
FULL ESTIMATED COST          0.22      0.22
```

FILE 'CAPLUS' ENTERED AT 18:49:46 ON 20 JUN 2009  
 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT.  
 PLEASE SEE "HELP USAGETERMS" FOR DETAILS.  
 COPYRIGHT (C) 2009 AMERICAN CHEMICAL SOCIETY (ACS)

Copyright of the articles to which records in this database refer is held by the publishers listed in the PUBLISHER (PB) field (available for records published or updated in Chemical Abstracts after December 26, 1996), unless otherwise indicated in the original publications. The CA Lexicon is the copyrighted intellectual property of the American Chemical Society and is provided to assist you in searching databases on STN. Any dissemination, distribution, copying, or storing of this information, without the prior written consent of CAS, is strictly prohibited.

FILE COVERS 1907 - 20 Jun 2009 VOL 150 ISS 26  
 FILE LAST UPDATED: 18 Jun 2009 (20090618/ED)  
 REVISED CLASS FIELDS (/NCL) LAST RELOADED: Apr 2009  
 USPTO MANUAL OF CLASSIFICATIONS THESAURUS ISSUE DATE: Feb 2009

CAPLUS now includes complete International Patent Classification (IPC) reclassification data for the third quarter of 2008.

CAS Information Use Policies apply and are available at:

<http://www.cas.org/legal/infopolicy.html>

This file contains CAS Registry Numbers for easy and accurate substance identification.

=> d his

(FILE 'HOME' ENTERED AT 18:49:04 ON 20 JUN 2009)

FILE 'CAPLUS' ENTERED AT 18:49:46 ON 20 JUN 2009

=> s hydrophilic platinum complex

109118 HYDROPHILIC

254558 PLATINUM

1490768 COMPLEX

L1 0 HYDROPHILIC PLATINUM COMPLEX  
 (HYDROPHILIC(W)PLATINUM(W)COMPLEX)

=> s platinum complex

254558 PLATINUM

1490768 COMPLEX

L2 9769 PLATINUM COMPLEX  
 (PLATINUM(W)COMPLEX)

=> s 12 and hydrophilic

109118 HYDROPHILIC

L3 28 L2 AND HYDROPHILIC

=> s 13 and lipid

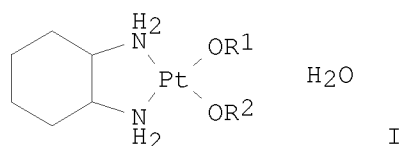
331776 LIPID

L4 2 L3 AND LIPID

=> d 1-2 bib abs

L4 ANSWER 1 OF 2 CAPLUS COPYRIGHT 2009 ACS on STN  
AN 1999:728076 CAPLUS  
DN 131:342017  
TI Lipid-soluble platinum(II) complex hydrates, and preparation of  
the hydrates and their freeze-dried preparations  
IN Tanno, Norihiko; Kishimoto, Hisakazu; Nakatsu, Michiko  
PA Sumitomo Pharmaceuticals Co., Ltd., Japan  
SO Jpn. Kokai Tokkyo Koho, 6 pp.  
CODEN: JKXXAF  
DT Patent  
LA Japanese  
FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
	-----	---	-----	-----	-----
PI	JP 11315088	A	19991116	JP 1999-57803	19990305
	JP 3007630	B2	20000207		
PRAI	JP 1998-73353	A	19980306		
OS	MARPAT 131:342017				
GI					



AB Title hydrates I (R1, R2 = C10-24 fatty acid residue) are prepared by crystallization of anhydrous I from mixts. of halogen-containing solvents, hydrophilic organic solvents, and H2O. I are dissolved into t-BuOH or its mixts. with halohydrocarbons and freeze-dried to give preps. useful as anticancer agents (no data).  
Dichloro[cyclohexane-(1R,2R)-diammine]platinum(II) (25 g) was treated with AgNO3 in H2O at 50-60° for 3 h, treated with myristic acid in CHCl3 in the presence of NaOH at 50° for 2 h, mixed with i-PrOH and H2O at 55°, and cooled to room temperature to give 42.42 g I (R1 = R2 = myristoyl), 3 g of which was dissolved into 750 mL t-BuOH in 25 min.

L4 ANSWER 2 OF 2 CAPLUS COPYRIGHT 2009 ACS on STN  
AN 1998:634913 CAPLUS  
DN 129:339460  
OREF 129:69005a  
TI Effects of new triphenylethylene platinum(II) complexes on the interaction with phosphatidylcholine liposomes  
AU Grenier, Guillaume; Berube, Gervais; Gicquaud, Claude  
CS Departement de Chimie Biologie, Universite du Quebec a Trois Rivieres, Trois Rivieres, QC, G9A 5H7, Can.  
SO Chemical & Pharmaceutical Bulletin (1998), 46(9), 1480-1483  
CODEN: CPBTAL; ISSN: 0009-2363  
PB Pharmaceutical Society of Japan  
DT Journal  
LA English  
AB In a previous work the authors synthesized a class of new antineoplastic drugs by coupling a cisplatin derivative to a triphenylethylene moiety similar to the antiestrogen, tamoxifen. These drugs differ in the number of hydroxy functions on the triphenylethylene rings and in the length of the linking

arm. To gain more insight into the cellular mechanism by which these new drugs act on cells, the authors studied, using differential scanning calorimetry, the effects of these compds. on the phase transition of membrane phospholipid (distearoyl phosphatidyl choline (DSPC)), and correlated these effects to drug cytotoxicity. The drugs without hydroxy function showed the highest cytotoxicity and induced little change on the thermogram of DSPC. Contrarily, the drugs bearing two or three hydroxy groups were less toxic, but induced important modifications of the thermogram. The authors suggest that the drugs with no hydroxy group enter the membrane, with the triphenylethylene moiety localized deep within the hydrophobic core of the bilayer and do not affect the cooperativity region (C2-C8). In contrast, drugs which bear hydroxy groups on the triphenylethylene rings system perturb the phospholipid mol. arrangement; this may be due either to the addnl. steric hindrance of the hydroxy functions in the core of the bilayer, or to their hydrophilic effect on the polar head of the lipid. In vitro, the cytotoxic effect of these drugs seems not to be related to their affinity for the estrogen receptor. Thus, the addition of a triphenylethylene moiety to the platinum(II) complexes increases the hydrophobicity, and consequently the resulting drugs become more permeable to the membrane, particularly the non-hydroxylated triphenylethylene derivs.

RE.CNT 18        THERE ARE 18 CITED REFERENCES AVAILABLE FOR THIS RECORD  
                  ALL CITATIONS AVAILABLE IN THE RE FORMAT

=> s l3 and nanoparticle#  
          141992 NANOPARTICLE#  
 L5            1 L3 AND NANOPARTICLE#

=> d bib abs

L5    ANSWER 1 OF 1    CAPLUS    COPYRIGHT 2009 ACS on STN  
 AN    2004:1088265    CAPLUS  
 DN    143:250075  
 TI    Use of self-assembling surfactants as templates and reactants for the  
       synthesis of noble metal particles  
 AU    Andersson, Martin; Harelding Ingelsten, Hanna; Palmqvist, Anders E. C.;  
       Skoglundh, Magnus; Holmberg, Krister  
 CS    Department of Applied Surface Chemistry, Chalmers University of  
       Technology, Goteborg, SE-412 96, Swed.  
 SO    Self-Assembly (2003), 105-111. Editor(s): Robinson, Brian H. Publisher:  
       IOS Press, Amsterdam, Neth.  
       CODEN: 69GGW5; ISBN: 1-58603-382-4  
 DT    Conference; General Review  
 LA    English  
 AB    A review. Recent work in the authors' laboratory regarding the use of the  
 title    surfactants as templates for noble metal particle synthesis is reviewed.  
       Nonionic surfactants containing polyoxyethylene chains can act both as  
       templates and as reducing agents in the synthesis of nanoparticles  
       of noble metals from a solution of the metal salt. In this paper, we show  
       that nanoparticles of platinum can be obtained by mixing one  
       microemulsion containing a water-soluble platinum complex,  
       [PtCl6]2-, with another microemulsion containing a reducing agent, such as  
       sodium borohydride (NaBH4). The choice of surfactant is decisive in  
       controlling the reaction rate. Whereas an alc. ethoxylate gives a fast  
       reaction regardless of the hydrophilic-lipophilic balance of the  
       surfactant, reaction in a microemulsion based on the anionic surfactant  
       sodium bis(2-ethylhexyl)sulfosuccinate (AOT) was relatively sluggish. The  
       difference is attributed to the nonionic surfactant assisting NaBH4 as  
       reducing agent. We also show that silver nanoparticles can be

produced by reduction of a silver nitrate solution with a nonionic surfactant, and a block copolymer of the polyoxyethylene-polyoxypropylene-polyoxyethylene type, without the use of any addnl. reducing agent. This reaction takes place in the narrow, channels of a reverse-hexagonal liquid-crystalline phase and the small silver particles became aligned into millimeter long fibers.

RE.CNT 19 THERE ARE 19 CITED REFERENCES AVAILABLE FOR THIS RECORD  
ALL CITATIONS AVAILABLE IN THE RE FORMAT

=> s solid lipid nanoparticle#  
1204996 SOLID  
331776 LIPID  
141992 NANOPARTICLE#  
L6 791 SOLID LIPID NANOPARTICLE#  
(SOLID(W)LIPID(W)NANOPARTICLE#)

=> s l6 and platinum complex  
254558 PLATINUM  
1490768 COMPLEX  
9769 PLATINUM COMPLEX  
(PLATINUM(W)COMPLEX)  
L7 0 L6 AND PLATINUM COMPLEX

=> s l6 and platinum  
254558 PLATINUM  
L8 2 L6 AND PLATINUM

=> d 1-2 bib abs

L8 ANSWER 1 OF 2 CAPLUS COPYRIGHT 2009 ACS on STN  
AN 2007:1207214 CAPLUS  
DN 149:95964  
TI Mosquito larva and pupa as models for screening anticancer activity  
IN Venkitachalam, Devarajan Padma; Shivajirao, Sonavane Ganeshchandra  
PA India  
SO Indian Pat. Appl., 25pp.  
CODEN: INXXBQ  
DT Patent  
LA English  
FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
	-----	----	-----	-----	-----
PI	IN 2006MU00210	A	20071019	IN 2006-MU210	20060214
PRAI	IN 2006-MU210		20060214		

AB The invention discloses a model for screening of anticancer drugs of synthetic, semisynthetic or natural origin. The invention more particularly discloses the use of mosquito larvae and mosquito pupa as model organisms for screening of drugs having anticancer activity. The invention is furthermore applied for screening of efficacy of drug delivery systems of anticancer drugs. Further, the invention concerns, in particular, a procedure for the identification and/or characterization of drugs having anticancer activity and extrapolates their potencies.

L8 ANSWER 2 OF 2 CAPLUS COPYRIGHT 2009 ACS on STN  
AN 2005:1075603 CAPLUS  
DN 143:373315  
TI Solid lipid nanoparticle formulations of  
platinum compounds  
IN Gasco, Maria Rosa; Gasco, Paolo; Bernareggi, Alberto  
PA Cell Therapeutics Europe S.r.l., Italy  
SO PCT Int. Appl., 20 pp.

CODEN: PIXXD2

DT Patent  
LA English  
FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2005092298	A1	20051006	WO 2005-EP3186	20050324
	W:	AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW			
	RW:	BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG			
	CA 2560900	A1	20051006	CA 2005-2560900	20050324
	EP 1734937	A1	20061227	EP 2005-716377	20050324
	R:	AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LI, LT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR			
	JP 2007530497	T	20071101	JP 2007-504372	20050324
	MX 2006010950	A	20070416	MX 2006-10950	20060925
	US 20080038371	A1	20080214	US 2007-594003	20070605
PRAI	US 2004-556754P	P	20040326		
	WO 2005-EP3186	W	20050324		

AB Solid Lipid Nanoparticles (SLN) of platinum compds., particularly of antitumor platinum complexes are disclosed. The nanoparticles of the invention are obtained by a process comprising: a) preparing a first microemulsion by mixing a molten lipid, a surfactant, and optionally a co-surfactant and the platinum compound aqueous solution; b) preparing a solution by mixing a surfactant and optionally a co-surfactant in water, heating to complete solution, preferably at the same melting temperature of the lipid used in a) and adding a co-surfactant; c) dispersing the microemulsion obtained in a) into the solution obtained in b) obtaining a multiple microemulsion c); d) dispersing the microemulsion obtained in c) in aqueous medium at a temperature ranging from 0.5°C to 4°C obtaining a dispersion of solid lipid microspheres; e) washing with aqueous medium through ultrafiltration the obtained lipid microspheres obtained in d) and lyophilizing, optionally in the presence of a bulking agent and of a cryoprotecting agent. SLNs of bis[trans-(diammine)(chloro)platinum (II)]- $\mu$ -(1,16-diamino-7,10-diazaheptadecane-N1,N16) dinitrate salt were prepared

RE.CNT 6 THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS RECORD  
ALL CITATIONS AVAILABLE IN THE RE FORMAT

=>

---Logging off of STN---

=>

Executing the logoff script...

=> LOG Y

COST IN U.S. DOLLARS	SINCE FILE	TOTAL
	ENTRY	SESSION
FULL ESTIMATED COST	54.36	54.58
DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)	SINCE FILE	TOTAL
	ENTRY	SESSION
CA SUBSCRIBER PRICE	-4.10	-4.10

STN INTERNATIONAL LOGOFF AT 18:59:31 ON 20 JUN 2009